

HYDROXYGRINDELANE DERIVATIVES BY MICROBIAL TRANSFORMATION

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Abstract—The microbial transformation of methyl 7 α ,8 α -epoxygrindelate and 6,8(17)-dehydrogrindelic acid by cultures of *Aspergillus niger* and *Penicillium brevicompactum* produced the corresponding 3 α -hydroxygrindelanes.

INTRODUCTION

In continuation of our effort to produce hydroxygrindelanes with potential insecticidal activity, we have subjected methyl 7 α ,8 α -epoxygrindelate and 6,8(17)-dehydrogrindelic acid to microbial transformation using *Aspergillus niger* X-172 and *Penicillium brevicompactum* 10418, respectively.

RESULTS AND DISCUSSION

We reported earlier [1] the microbial transformation of grindelic acid (3) into 3 α -hydroxygrindelic acid (5) in good yield. We now report that methyl 7 α ,8 α -epoxygrindelate (1) and 6,8(17)-dehydrogrindelic acid (7) are transformed analogously into methyl 3 α -hydroxy-7 α ,8 α -epoxygrindelate (2) and 3 α -hydroxy-6,17-dehydrogrindelic acid (9), respectively.

In the EIMS, 2 displayed an M^+ peak at m/z 366 (m/z 366.2400 by HRMS: $C_{21}H_{34}O_5$ requires 366.2406), 16 mass units higher than that of 1, and a mass spectral pattern very similar to that of 1. This increment in the M^+ of 16 mu together with the presence of minor but diagnostic peaks at m/z 348 [$M - H_2O$]⁺, 293 [$M - CH_2COOMe$]⁺, 275 [m/z 293 - H_2O]⁺, 257 [m/z 293 - 2 H_2O]⁺, and a strong hydroxyl band (3450 cm^{-1}) in the IR (neat) spectrum of 2 clearly suggested that 1 had been hydroxylated. Since the m/z 226 species, which occurs as the base peak in 1 and 2, includes the THF and B ring portions but lacks the A ring (Scheme 1), the hydroxyl group appeared to be in ring A. Supporting evidence came from a peak of appreciable intensity at m/z 252 (not observed in 1), shifted from m/z 236 in 1 as required by the substitution of a hydroxyl group in ring A of 2: its possible formulation is outlined in Scheme 1 (composition verified by HRMS). Since the peaks in the low mass region (below m/z 226) of the EIMS of 1 and 2 occur with same masses, and the same microorganism (*A. niger*) was used, the C-3 locus was favoured for the hydroxyl function in 2.

The ^1H (Table 1) and the ^{13}C (Table 2) NMR spectra confirmed the hydroxyl group in 2 to be in the 3 position,

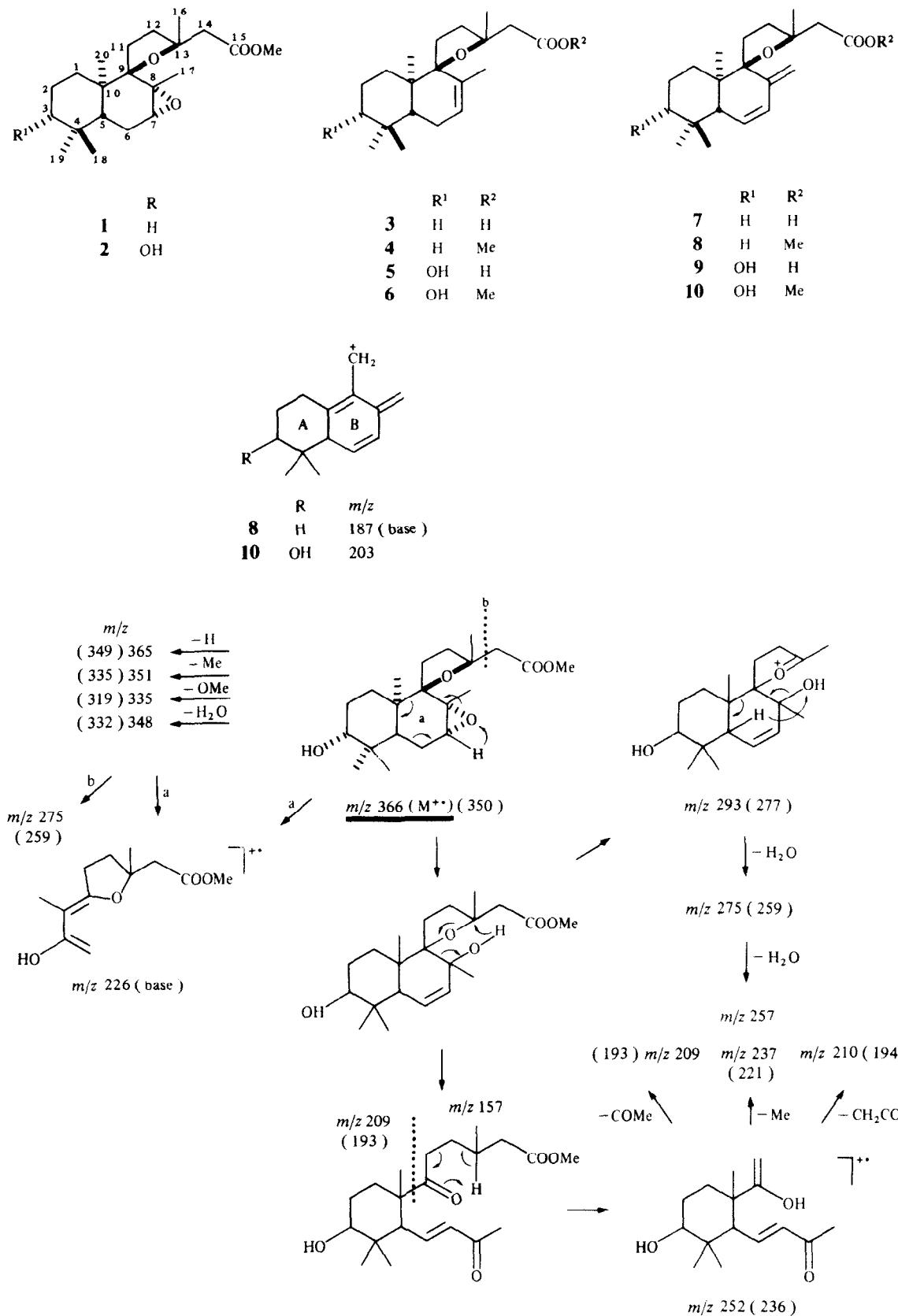
and showed it to be equatorial. The C-3 proton absorbs like that in 5 [1], and must be axial to have a coupling constant as large as 11.0 Hz. Compound 1 matches the NMR spectra of 2 except for the A ring peaks, which more closely match those of 5.

The microbial transformation of 7 into its corresponding 3 α -hydroxy derivative (9) was done on a mixture containing 75% of 3 and 25% of 7 (GC); we did not separate these substances from one another since they have virtually identical R_f values on the chromatographic solvent systems used. Neither the biotransformation products 5 and 9 nor their corresponding methyl ester derivatives 6 and 10 were separated from one another for the same reason but their structures were evident from the ^1H NMR spectra of the 3:1 mixtures (Table 1) and received support from the EIMS of the mixture of 6 and 10. Like diene 8, the mixture exhibited little fragmentation but the peak at m/z 277 (loss of $\text{CH}_2\text{CO}_2\text{Me}$) and a retro-Diels-Alder fragment at m/z 210 [base, corresponding to m/z 196 (base) in 5], both derived from M^+ (m/z 350, not observed) of 6 and diagnostic peaks at m/z 348 (M^+ , 3.2%) and 203 (19.4%), both shifted from m/z 332 (M^+) and 187 (base) in 8, clearly suggested that both 6 and 10 were hydroxylated. Since the m/z 187 ion (Fig. 1) in 8 includes the A and B rings, the hydroxyl group in 10 must be in ring A.

EXPERIMENTAL

See references [2] for the analytical procedures, [1] for the description of microorganism and fermentation procedure and [3] for the GC analysis. A sample of methyl 7 α ,8 α -epoxygrindelate (1) and of grindelic acid (3)-6,8(17)-dehydrogrindelic acid (7) mixture (3:1) were obtained by previously published procedure [4].

Methyl 3 α -hydroxy-7 α ,8 α -epoxygrindelate (2). The *A. niger* fermentation product (100 mg) was separated into Et_2O soluble and insoluble fractions. From the Et_2O -soluble fr. (78 mg), 2



Scheme 1. Diagnostic EIMS fragments of **2**. Figures in parentheses represent analogous ions in the EIMS of **1**.

Table 2. ^{13}C NMR spectral data for compounds **1**, **2** and **5** (22.63 MHz, CDCl_3 , with TMS as int. standard)

C	1[6]	2	5[1]
1	31.8	29.7	30.8
2	18.3	27.2	27.0
3	41.7	78.1	78.5
4	33.1	38.6	38.8
5	37.2	37.2	42.3
6	22.9	22.5	23.8
7	61.5	61.9	128.4
8	59.0	59.0	133.2
9	88.3	88.1	92.1
10	39.8	39.6	40.5
11	29.1	29.2	27.7
12	37.5	37.4	39.2
13	81.4	81.5	81.2
14	47.1	47.0	47.4
15	171.9	171.8	172.4
16	26.7	27.2	26.8
17	23.1	22.9	21.0
18	32.5	27.2	27.6
19	22.5	15.4	15.1
20	16.5	16.6	16.8
OMe	51.2	51.3	—

(40 mg, single spot by TLC) was isolated by PLC [*n*-hexane-EtOAc-AcOH (20:20:3), single development]. Its IR (text), ^1H NMR (Table 1), ^{13}C NMR (Table 2) and mass (Scheme-1) spectra were in accord with the structure shown.

Methyl 3α -hydroxy-6,8(17)-dehydrogrindelate (**10**) and methyl 3α -hydroxy-grindelate (**6**). The *P. brevicompactum* fermentation product (240 mg) was extracted with Et_2O and filtered. From the Et_2O -soluble filtrate (185 mg), the major spot (65.8 mg) was isolated by PLC [petrol- Et_2O -AcOH (5:5:1), single development] and methylated [5]. The methylated product (53 mg), when submitted to PLC [petrol- Et_2O - EtOAc (15:2:8), two developments], gave a TLC single spot fr (24.3 mg), shown by ^1H NMR to be a mixture of **6** (70%) and **10** (30%). Its IR (text), ^1H NMR (Table 1) and mass (text) spectra were in accord with the structures shown.

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*Coupling constants in Hz in parenthesis.

¹⁷ May be interchanged

May be overlooked.
Obscured by peaks due to an impurity.